

B-5

USSR/ Physical Chemistry - Crystals

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 10929

analogous to polyhedrons in structures of  $\text{PtSn}_4$ ,  $\text{PdPb}_4$ ,  $\text{AuSn}_4$ ,  $\text{CuAl}_2$ ,  $\text{CoGe}_2$ . Bi atoms have coordination number 11 (9 atoms Bi and 2 atoms Rh). Interatomic distance Bi-Rh is 2.80 Å, and is less by 3% than sum of atomic radii. Distance Bi-Bi is from 2.80 to 3.50 Å; three of them are close to least distance, and six to greatest distance in pure Bi. The question concerning analogy in structure of alpha-Bi<sub>4</sub>Rh and garnet  $\text{Ca}_2\text{Al}_2\text{Si}_3\text{O}_12$  is discussed.

Card 2/2

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CIA-RDP86-00513R002064620006-5"

ZHDANOV, G. S.

AUTHOR  
TITLE  
PERIODICAL

KINCHIN, G.H., PEASE, R.S.

PA - 2040

The Displacement of Atoms in Solids under the Effect of Radiation. (Russian)  
Uspekhi Fizicheskikh Nauk, 1956, Vol 60, Nr 4, pp 590-615 (U.S.S.R.)

Received 1/1957

Reviewed 3/1957

ABSTRACT

This is the Russian translation of articles 2 and 3 of the survey by the above mentioned authors (under the editorship of G.S.ZHDANOV), Reports on Progress in Physics, 18, 1 (1955), this survey is arranged as follows.  
I. The production of displaced atoms. 1) energetical deliberations. The energy of displacement. (It is to be expected that an energy that is higher than 5 eV is needed for the displacement of an atom from its position in the lattice of a solid, the threshold values of radiation energy are given in a table for the displacement energy of 25 eV which is assumed for all further cases). 2) The number of displaced atoms. The moved atoms, bases of the theory, collisions of solid spheres, Rutherford's collisions, moved atoms which are slowed down to a full stop, fast neutrons, electrons, gamma rays. 3) The distribution of the displaced atoms and vacancies. a) The models based upon the investigation of collisions. b) "displacement wedges" and "thermal wedges". 4) Effects in compounds. Collisions that lead to the displacement of atoms, dissolution of order.  
II. Restoration of defects. In most cases the physical properties return to their original values if the temperature of the sample is increased after irradiation. As a rule it may be expected that also in the case of irradiation a certain restoration of defects takes place by heating, usually in several stages with different values of activation energy in

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PA - 2040

The Displacement of Atoms in Solids under the Effect of Radiation.

the various temperature intervals. The course taken by the restoration process may be described by the equation  $dn/dt = -cn\gamma \exp(-\epsilon/kT)$ . Here "n" denotes the number of the defects participating in the restoration process,  $\epsilon$  the activation energy,  $\gamma$  the so-called "order of reaction". The II. section is arranged as follows. 1) Recombination of impressed atoms and vacancies. Pairs of atoms and vacancies located close to each other, the disordered distribution of impressed atoms and vacancies. 2) Additional processes 3) the accumulation of destructions 4) saturation 5) Annealing by radiation.

ASSOCIATION  
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Not given  
Library of Congress

Card 2/2

ZHDANOV, G. S. and VLASENKO, V. I.

The National Committee for Crystallography of the USSR, Moscow-

"Automatic Synthesis of Two-Dimensional Crystal Structure Patterns"  
(Section 1(1)-8, a paper submitted at the General Assembly and International  
Congress of Crystallography, 10-19 Jul 57, Montreal, Canada.

C-3,800,89

ZHDANOV, G. S. (Prof.)

"The Work of the Electronic Computing Machine "Kristall" and demonstrated it."

report presented at Scientific Conference at the Inst. for Physical Chemistry  
imeni L. Ya. Karpov, Acad. Sci. USSR, Nov 1957.

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CIA-RDP86-00513R002064620006-5"

ZHDANOV G.S.  
VLASENKO, V.I.; ZHDANOV, G.S.; SOKOLOV, A.D.

Dynamic method for analog representation of two-dimensional  
functions. Priborostroenie no.6:11-14 1957. (MIRA 10:7)  
(Electronic calculating machines) (Mathematical models)

"APPROVED FOR RELEASE: 07/19/2001 CIA-RDP86-00513R002064620006-5

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REF ID: A6513

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CIA-RDP86-00513R002064620006-5"

ZHDANOV, G.S.

AUTHOR: Ozerov, R.P., Gol'der, G.A. and Zhdanov, G.S. 70-2-3/24

TITLE: An X-ray structural investigation of the oxygen vanadium bronzes of sodium and potassium  $Na_{0.33}V_2O_5$ . (Rentgenograficheskoye issledovaniye strukturny kislorodnykh vanadiyevykh bronzy natriya i kaliya  $Na_{0.33}V_2O_5$ )

PERIODICAL: "Kristallografiya" (Crystallography), 1957, Vol. 2, No. 2, pp. 217-225 (U.S.S.R.)

ABSTRACT: Experimental. The valency state of V in bronzes and in vanadium-sulphur-oxygen catalysts is particularly of interest. Crystals of composition  $(K, Na)_2O \cdot V_2O_4 \cdot 5V_2O_5$  were obtained as black laths having a blue metallic lustre. They showed a large number of faces including the simple forms 102, 101, 100, 001 variously developed. X-ray photographs assigned them to the Laue class  $2/m = C_{2h}$ . Weissenberg and oscillation photographs (11.456 cm diameter camera) with Fe radiation gave unit cell dimensions  $a = 10.039$ ,  $b = 3.605$ ,  $c = 15.335$  Å (all  $\pm 0.003$  Å) and  $\beta = 109^{\circ}12' \pm 3'$ , for the sodium compound  $Na_{0.33}V_2O_5$ . This gives  $V = 524.2$  Å<sup>3</sup>. The compound  $K_2V_2O_3$  had  $d_{obs.} = 3.57$  g/cm<sup>3</sup> making  $z = 1$  (0.97).  $d_{calc.}$  is then 3.60.

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70-2-3/24

An X-ray structural investigation of the oxygen vanadium  
bronzes of sodium and potassium  $Me^{0.33}V_2O_5$ . (Cont.)

The possible space groups (from the extinctions) were A2/m, A/2 and Am. On the basis of a knowledge of the crystal chemistry of the oxides of V, Mo and W and of the W bronzes the group A2/m was chosen. This is confirmed by the dimension  $b$  which leads to the expectation of octahedra or trigonal bipyramids (see R.P. Ozerov - Usp. Khim. 24, 951, 1955). Using Mo radiation 800 reflections were measured from retigraph pictures using comparison scales. No extinction corrections were applied.  $P(x, z)$  was constructed. A.D. Wadsley's determination of the structure of  $Na_{2-x}V_6O_{15}$  helped in solving this Patterson projection. Projections for both Na and K bronzes were constructed. Several atoms overlap and this was one reason for repeating Wadsley's work. The Fourier section at  $y = 0$  was calculated giving co-ordinates for the K bronze very close to those found by Wadsley (Acta Crystallography, 8, 695, 1955) for the Na bronze. A table of interatomic distances is given. Slight differences naturally occur in the Me-O distances (as observed for K bronze first followed by Wadsley's value for the Na bronze); Me-O<sub>8</sub> (2.50, 2.46); Me-O<sub>8</sub> (2.49, 2.75); Me-O<sub>4</sub> (2.56, 2.51);

Card 2/3

70-2-3/24

An X-ray structural investigation of the oxygen vanadium bronzes of sodium and potassium Me<sub>0.33</sub>V<sub>2</sub>O<sub>5</sub>. (Cont.)

Me-O<sub>6</sub> (2.57, 2.51); Me-Me (1.68, 2.22). The geometry of the structure is discussed. The structure is built from strongly distorted VO<sub>6</sub> octahedra. The distortion is so great that certain groupings are better regarded as trigonal bipyramids. The polyhedra differ greatly among themselves V-O distances oscillating to 1.55, 1.79, 1.89, 2.00 and 2.68 Å. There is a strong correspondence with the structure of the V oxides. The alkali atoms lie in canals between the octahedra each surrounded by seven oxygens. Seven-fold co-ordination is rare but is also found in the ion (NaF<sub>7</sub>)<sup>-2</sup> and in Bi<sub>2</sub>Ni.

Card 3/3 There are 5 figures, 2 tables and 20 references, 10 of which are Slavic.

ASSOCIATION: Ya.V. Samoylov Scientific Institute for Fertilisers and Insecto-fungicides. (Nauchnyy Institut po Udobreniyam i Insektofungisidam im Ya.V. Samoylova)

SUBMITTED: September 21, 1956.

AVAILABLE: Library of Congress

ZHDANOV, G.S.

AUTHOR: Zhdanov, G.S., Zhuravlev, N.N., Stepanova, A.A. and  
Umanskiy, M.M. 70-2-16/24

TITLE: The crystal chemistry of metal hexaborides. (Kristallo-  
khimiya geksaboridov metallov)

PERIODICAL: "Kristallografiya" (Crystallography), 1957, Vol.2,  
No.2, pp. 289-290 (U.S.S.R.)

ABSTRACT: The  $MeB_6$  structure consists of a three-dimensional frame-work of B atoms with Me atoms in the interstices and is formally like the CsCl structure with  $B_6$  and Me units. In group II, Be, Mg, Ca, Sr and Ba form hexaborides as do Y, La, Ce, Pr, Nd, Gd, Er, Yb in group III. The formation by the remaining lanthanides and Sc of hexaborides can also be predicted. In group IV,  $SiB_6$  and  $ThB_6$  are known. When the unit cell sizes are plotted against atomic radii the compounds fall into three series corresponding to the three groups. Coefficients of thermal expansion of  $CaB_6$  and  $SiB_6$  have been measured by X-ray diffraction (5.9 and  $6.5 \times 10^{-6}$ , respectively and are added to the measurements already made (A.A. Stepanova and M.M. Umanskiy, Trudy soveshchaniya po khimii bora i ego soedineniy. Fiz.-Khim. int. im. Karpova, 1955) for Ce, La and Ba ( $7.3$ ,  $6.4$  and  $6.8 \times 10^{-6}$ , respectively). The coefficient

Card 1/2

C

AUTHOR: Zhdanov, G.S.

Vlasenko, V.I. and Zhdanov, G.S.

70-3-6/20

TITLE: Automatic synthesis of two-dimensional crystal structure patterns (Avtomatushkiy sintez dvumernykh izobrazheniy atomykh struktur)

PERIODICAL: "Kristallografiya" (Crystallography), 1957,  
Vol. 2, No. 3, pp. 358 - 365 (U.S.S.R.)

ABSTRACT: By means of high speed digital computers it is possible to summate Fourier series and thus compute numerical tables within a short time. However, digital computers are unsuitable for further analysis; for this purpose, an automatic machine is required. The first problem is to convert the numerical tables into a more readily usable form.

The simplest method of this synthesis is the construction of a mosaic image. In this case each number in the table is replaced by a round or a square spot with the colour corresponding to this number. The whole of the colour spots gives a visual mosaic image of the electron density. This is due to the ability of the eye to integrate discrete elements in smooth forms.

Another method is that of the model section; it is a better method but a more difficult one. In this case it is necessary to construct a model of the two-dimensional function

Card 1/3

Automatic synthesis of two-dimensional crystal structure patterns. (Cont.)

70-3-6/20

of the electron density, then to dissect the model with a series of planes parallel to XY on different levels and to register on a flat screen the intersection lines. The model is constructed by two-dimensional interpolation first along the X axis, and then along the Y axis. The results of the first (X-axis) interpolation is recorded in a special storage device and represents the initial data for the second (Y axis) interpolation.

The first interpolation is performed by electronic circuits, the results being recorded in parallel circular tracks on a rotating magnetic drum. The second (Y axis) interpolation of these results, along the drum axis, as well as all other operations are also made by the electronic circuits.

The results of the second interpolation can be considered as being a curve, representing the distribution of electronic density along the Y axis (drum axis). An amplitude discriminator automatically dissects this curve parallel to the Y axis by a series of straight lines, representing the given set of electron-density values. The intersection points are recorded as light spots on a CRT screen whereby the coordinates of these spots correspond to the position of the

card 2/3

70-3-6/20

Automatic synthesis of two-dimensional crystal structure patterns. (Cont.)

intersection points. Thus, during one revolution of the magnetic drum the light spots form dotted lines, representing the whole electron-density map.

The use of the magnetic drum and electronic circuits for the above mentioned purposes allows obtaining an electron-density map on a CRT screen within a few seconds after ending the summation of the Fourier series on the high-speed computer.

There are 8 figures and 7 references, 2 of which are Slavic.

ASSOCIATION: Physico-chemical Institute im. I.Ya.Karpov.  
(Fiziko-khimicheskiy Institut im. I.Ya. Karpova)

SUBMITTED: February 22, 1957.

AVAILABLE: Library of Congress  
Card 3/3

AUTHORS: Zhdanov, G.S., Solov'yev, S.P. and Venevtsev, Yu.N. 70-5-12/31  
 TITLE: The Structural Coefficients of the Internal Field in Ferro-electrics with the Perovskite-type Structure. (Strukturnyye koeffitsiyenty vnutrennego polya v segnetoelektrikakh so strukturoy tipa perovskita)

PERIODICAL: Kristallografiya, 1957, Vol.2, No.5, pp. 639-645 (USSR)

ABSTRACT: Data published in the literature for the values of the structural coefficients of the internal field in perovskite-type crystals are critically examined. Relations are set up between the idealised cubic perovskite and the cases in which there are dipoles in the [001], [011] or [111] directions. The structural coefficients in the [001], [011] or [111] directions. The of  $\text{PbTiO}_3$  at room temperature taking account of the ionic displacements.

The field at the i-th ion is:

$$E_i = E + \sum_{k=1}^m \left( \frac{4\pi}{3} + C_{ik} \right) p_k$$

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where  $m$  is the number of sub-lattices each consisting of the ions of the k-th sort,  $E$  is the external field,  $p_k$  is the

70-5-12/31

The Structural Coefficients of the Internal Field in Ferroelectrics  
with the Perovskite-type Structure.

dipole moment of the k-type of ions,  $C_{ik}$  are the structural coefficients. In the case where the dipoles are in the z-direction:

$$C_{ik} = \sum_j \frac{2z_{jk}^2 - x_{jk}^2 - y_{jk}^2}{(x_{jk}^2 + y_{jk}^2 + z_{jk}^2)^{5/2}}$$

where  $x_{jk}$ ,  $y_{jk}$ ,  $z_{jk}$  are the co-ordinates of the j-th dipole of the k-th sort relative to a dipole of the i-th sort and summation is over each dipole of the k-th sort. For the cubic cell, the  $C_{ik}$  can be expressed in terms of two quantities P and Q. The values which various authors find for these values are compared, the best values being  $P = -15.04102/V$  and  $Q = 4.33387/V$  as found by McKeehan (Phys. Rev. 42, 913, 1933 and 72, 78, 1947).

Card2/4

70-5-12/31

The Structural Coefficients of the Internal Field in Ferroelectrics  
with the Perovskite-type Structure.

$$c_{ik} = \begin{matrix} 0 & 0 & -2Q & Q & Q \\ 0 & 0 & -2P & P & P \\ -2 & Q & -2P & 0 & Q \\ Q & P & Q & 0 & -2Q \\ Q & P & Q & -2Q & 0 \end{matrix}$$

There are 5 sub-lattices for the  $\text{ABO}_3$  formula.  
 $c_{ik}$  are the values for the truly cubic cell.

Where the distortions of the cubic cell are small (1%) the  $c_{ik}$  are different from the  $c_{ik}$  only by 2-3%. Even for  $\text{PbTiO}_3$ , where the distortions are large, these do not differ by more than 20%. The actual values of the coefficients for tetragonal  $\text{PbTiO}_3$  where  $c/a = 1.064$  are calculated and tabulated with those of McKeehan (cubic, by Ewald's method) and of Hagendorf ( $\text{BaTiO}_3$  with  $c/a = 1.010$ ) (Zeit.f.Physik, 132, 394-421, 1952). There are 1 figure, 3 tables and 14 references, 2 of which are Slavic.

ASSOCIATION: Karpov Physico-chemical Institute  
(Fiziko-khimicheskiy Institut im. L.Ya. Karpova)  
Card 3/4

70-5-12/31

The Structural Coefficients of the Internal Field in Ferroelectrics  
with the Perovskite-type Structure.

SUBMITTED: May 31, 1957.

AVAILABLE: Library of Congress

Card 4/4

ZHDANOV, G.S.

MEPEL'BAUM, V.A.; SEVAST'YANOV, N.G.; GUREVICH, M.A.; ORMONT, B.Y.; ZHDANOV,  
G.S.

Phases formed in the system chromium -- boron. Part 1: Formation  
of "β-chromium" under the influence of small additions of boron.  
Zhur. neorg. khim. 2 no.8:1848-1854 Ag '57. (MIRA 11:3)  
(Chromium) (Boron)

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ZHDANOV, G.S., professor.

~~Scientific works by the Department of Solid State Physics. Vest.Mosk.~~  
un. 12 no.1:195-196 '57. (MLRA 10:8)  
(Solids)

APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R002064620006-5"

ZHDANOV, G.S.

Atomic structure and chemical bonding. Vest. Mosk.un.Ser.mat.  
mekh.astron. fiz. khim. 12 no.4:61-78 '57. (MIRA 11:5)

1.Kafedra fiziki tverdogo tela Moskovskogo gosudarstvennogo  
universiteta. (Chemical bonds) (Atoms)

ZHDANOV, G.S.

AUTHORS:

Venevtsev, Yu.N. and Zhdanov, G.S.

TITLE:

Crystalliochemistry of Ferroelectrics of Perovskite Structure. (Kristallokhimiya segnetoelektrikov so strukturoy tipa perovskita)

PERIODICAL:

Izvestiya Akademii Nauk SSSR, Vol. XXI, #2, pp 275  
-285, 1957, USSR, Seriya fizicheskaya

ABSTRACT:

The data available on some ferroelectrics and anti-ferroelectrics make it presently possible to classify these compounds by their structural properties and to determine relations between the structure and the character of spontaneous elec polarization.

For crystallic structure of ferroelectrics with  $ABO_3$  composition of the perovskite type, the existence of  $BO_6$  octahedrals joined by their vertices is a characteristic feature; empty gaps between them are occupied by A-type ions.

The structure of the perovskite type depends mainly up on the ratios of radii of constituent ions. The valence of A ions may be 1,2,3 and that of B ions

Card 1/5

TITLE:

Crystalllochemistry of Ferroelectrics of Perovskite Structure. (Kristallokhimiya segnetoelektrikov so strukturoy tipa perovskita)

- 5,4,3 respectively. When A ions and oxygen ions have equal radii, these ions form a densely packed cubic structure. Within the oxygen octahedrals of this structure, B-ions may possess a maximum radius equal to 0.414 of the oxygen ion radius, i.e. 0.56 Å. Thus an ideal contact of adjacent ions takes place when the ratio  $t = \frac{r_A + r_B}{\sqrt{2}(r_A + r_B)}$  is equal to 1.

According to Megaw (8) and Naray-Szabo (9) it is sufficient to take into account co-ordination numbers of ions for evaluating the factor t by the formula:

$$t = \frac{r_{A(12)} + r_O}{\sqrt{2}(r_A + r_O)}$$

where  $r_A$ ,  $r_B$  and  $r_O$  are tabular values of A,B,O ion radii, and subscript (12) means tabular value of the A ion radius corrected for the case of co-ordination number 12.

Card 2/5

TITLE:

Crystalliochemistry of Ferroelectrics of Perovskite Structure. (Kristallokhimiya segnetoelektrikov so strukturoy tipa perovskita)

Peculiar properties of BaTiO<sub>3</sub> are connected with the fact that the titanium ion has a "free" space in the BaTiO<sub>3</sub> cell. The main condition for the ferroactivity of an ion is that free space in the cell must be available.

Classification results of ferroelectrics and antiferroelectrics of the BaTiO<sub>3</sub> group are presented in Table 1. Inspection of this table shows a definite regularity between the t-value and elec polarization character. If t has a value considerably less than 1, the compound has antiferroelectric properties.

Ferroactive cations (A or B) are displaced at a certain temperature (lower than Curie point) from their symmetric positions and thereby bring about the polar rebuilding of the whole cell.

Card 3/5

TITLE:

Crystallochemistry of Ferroelectrics of Perovskite Structure. (Kristallokhimiya segnetoslektrikov so strukturoy tipa perovskits)

Displacements of B and A ferroactive cations are observed along the axes of the 2nd and 3rd and 4th orders, which result in monoclinic, rhombohedral and tetragonal distortions respectively. In the cells of antiferroelectrics, antiparallel displacements of ferroactive A cations along the axis of the 2nd order are observed.

Co-ordination numbers of A and B ferroactive cations characterizing their displacements along various axes are given in Table 2.

Ferroelectrics with  $t > 1$  and accompanied by temperature changes perform 3-phase transitions. Ferroelectric  $\text{PbTiO}_3$ , with  $t < 1$  and lowering of the temperature performs only a one-phase transition. None of the known ferroelectrics and antiferroelectrics with  $t \leq 1$  has shown, thus far subsequent displacements of the A

Card 4/5

TITLE:

Crystallochemistry of Ferroelectrics of Perovskite  
Structure. (Kristallokhimiya segnetoelektrikov so  
strukturoy tipa perovskita)

cation along the 3 possible directions of displacements.

Ferroelectrics possessing the perovskite structure are  
compounds with principally ionic character of bonds.

The article given 3 figures and 2 tables. The bibli-  
ography contains 46 references, of which 10 are Slavic  
and 1 Hungarian.

INSTITUTION: Physico-Chemical Institute imeni L.Ya. Karpov

PRESENTED BY:

SUBMITTED: No date

AVAILABLE: At the Library of Congress.

Card 5/5

ZHDANOV, G.S.

5(2)

PHASE I BOOK EXPLOITATION

SOV/1916

Vsesoyuznoye soveshchaniye po khimii bora, 1955

Bor: trudy Konferentsii po khimii bora i yego soyedineniy (Boron; Transactions of the Conference on the Chemistry of Boron and Its Compounds) Moscow, Goskhimizdat, 1958. 189 p. Errata slip inserted. 2,400 copies printed.

Ed.: G.P. Luchinskiy; Tech. Ed.: M.S. Lur'ye.

PURPOSE: This book is intended for chemists, as well as for industrial personnel working with boron and its compounds.

COVERAGE: This collection contains 24 studies on the chemistry, crystalline structure, physicochemical properties, and technology of boron and its compounds. Twenty-two of the studies were presented at the All-Union Conference on Boron Chemistry, held at the Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Research Physicochemical Institute im. L. Ya. Karpov) in

Card 1/6

Boron; Transactions of the Conference (Cont.) Sov/1916

December 1955. Two of these articles deal with the thermo-chemistry of boron. The two studies on "borundum" production are being published for the first time. The studies are well illustrated and accompanied by bibliographies.

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Boran; Transactions of the Conference (Cont.) SOV/1916

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Ormont, B.F., V.A. Epel'baum, and I.G. Shafran. An  
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and in Testing Its Properties 182

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Card 6/6

AUTHOR:

Preobrazhenskiy, Yu.A., Zhdanov, G.S.

113-58-7-2/25

TITLE:

The Economy of Smelt-Model Casting (Ob ekonomike lit'ya po vyplavlyayemym modelyam)

PERIODICAL:

Avtomobil'naya promyshlennost', 1958, Nr 7, pp 4-5 (USSR)

ABSTRACT:

In 1956, NIITAvtoprom established a catalogue of 500 individual parts used in automobile, motorcycle and bicycle production. Reduction of this list to 175 parts is possible by a very accurate method of evaluation. First, the parts are grouped by weight, intricacy and design, and the possibility of reducing the amount of mechanical machining is considered. Accurate casting to desired shape must then be effected wherever it is possible. The economy obtained by smelt-model casting of rocker arm yokes is demonstrated in a table. In the automobile building industry, this economy, by the smelt-model casting process, applies to forgings and rolled iron parts of up to 350 grams, which are later subjected to machining by cutting. In most cases, a transfer of parts made of machined rods to automats or semi-automats does not pay. The introduction of shell casting in the Moskovskiy zavod malolitrazhnykh avtomobiley (Moscow Light Car Plant) has resulted in a reduction of operators and a diminished consumption of electric

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The Economy of Smelt-Model Casting

113-58-7-2/25

power. There is 1 table.

ASSOCIATION: NIITAvtoprom (NIITAvtoprom)

1. Metals—Casting 2. Castings—Economic aspects

Card 2/2

70-3-2-1/26

AUTHORS: Vlasenko, V.I. and Zhdanov, G.S.

TITLE: Optical Methods of Summing Fourier Series (Opticheskiye metody summirovaniya ryadov Fur'ye)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 2, pp 135 - 140 (USSR)

ABSTRACT: The various optical methods of summing Fourier series are classified according to their characteristics: sequential simultaneous, one mask/ set of masks, white/ coloured light, cinematographic/static. The properties of the photographic materials used in the Bragg-Huggins masks and in the von Eller photosommateur are discussed with reference to range of linearity and to maximum density. A new method (possibly not yet realised) is described. The Huggins masks, translated to give correct phases, pass in turn before a lamp (presumably modulated) in rapid succession so that all merge, owing to the persistence of vision, to give a summation. Cinema technique is required for this apparatus but it uses only one projecting lens and the resulting summation can be very easily recorded photographically. A machine where each mask is projected simultaneously onto a white screen and the resulting pattern of weak or strong illumination/represents the summation is also described, but appears rather impracticable. The von Eller machine is commended.

Optical Methods of Summing Fourier Series

70-3-2-1/26

There are 4 figures and 12 references, 5 of which are Soviet,  
4 French and 3 English.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova  
(Physico-chemical Institute, Im. L. Ya. Karpov)

SUBMITTED: February 1, 1957  
Card 2/2

AUTHORS: Verbitskaya, T.N., Zhdanov, G.S., Venevtsev, Yu.N.  
and Solov'yev, S.P. 70-3-2-9/25

TITLE: Electrical and X-ray Investigations of the System  
 $\text{BaTiO}_3 - \text{BaZrO}_3$  (Elektricheskiye i rentgenograficheskiye  
issledovaniya sistemy  $\text{BaTiO}_3 - \text{BaZrO}_3$ )

PERIODICAL: Kristallografiya, 1958, Vol. 3, Nr 2, pp 186 - 196  
(USSR).

ABSTRACT: Various solid solutions of  $\text{BaTiO}_3 - \text{BaZrO}_3$  were prepared, having up to 30 mol% of the latter, by heating appropriate mixtures of  $\text{BaCO}_3$ ,  $\text{TiO}_2$  and  $\text{ZrO}_2$  at  $1400 \pm 10^\circ\text{C}$ . The resulting materials were examined by the Debye-Scherrer method with a camera of diameter 11.4 cm and Cr or Cu radiation. With Cr radiation the lines 310 and 222 occur at a sufficiently high angle to give accurate cell dimensions. (For Cu radiation the appropriate lines are 501 and 431) The significance of the splitting of the lines under the distortions observed is explained. Dimensional measurements were made to  $\pm 0.001 \text{ \AA}$ , monoclinic angle to  $\pm 1.5'$ , rhombohedral angle to  $\pm 1'$  and axial ratio to  $\pm 0.0005$ . Card 1/3 For the pure compounds the cell dimensions were found to be:-

Electrical and X-ray Investigations of the System  $\text{BaTiO}_3 - \text{BaZrO}_3$  70-3-2-9/26

$\text{BaTiO}_3$ ,  $a = 3.990$ ,  $c = 4.027$ ,  $c/a = 1.0093$ ,  $V = 64.12$ ; and  $\text{BaZrO}_3$ ,  $a = 4.186$  and  $V = 73.35$ . From 0 to 2 mol% of zirconate the dimensions of the tetragonal phase approached each other slightly. From 2 to 6.5% the solid solution was pseudo-monoclinic with  $a = c$  and the monoclinic angle decreasing from  $90^\circ 08.5'$  to  $90^\circ 04.0'$ . From 6.5 to 20 mol%, the solution was rhombohedral with the rhombohedral angle equal to  $89^\circ 57'$  and increasing from  $4.0177$  to  $4.0440$ . Above 20% the solution was cubic with an increasing from  $4.0447$  to  $4.0616$  at 30%. Over the whole range the volume of the unit cell increased linearly from  $64.12$  to  $67.00 \text{ \AA}^3$  with no breaks at the phase transitions. In pure  $\text{BaTiO}_3$  three phase transitions (all with hysteresis) are observed on changing its temperature. They are at  $120^\circ$ ,  $0-5^\circ$  and  $-70$  to  $-80^\circ\text{C}$ . These transition points were measured for a range of compositions. Below 10% zirconate all four phases occur at appropriate temperatures, between 10 and 15% there is a confused region and above 15% only the cubic and rhombohedral phases occur. Measurements were also made of the dielectric constant of the material at various temperatures

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70-3-2-9/26

Electrical and X-ray Investigations of the System  $\text{BaTiO}_3 - \text{BaZrO}_3$

and compositions.

The phase diagram constructed is like that found for  $\text{BaTiO}_3 - \text{BaSnO}_3$  by Smolenskiy and Isupov (DAN, 1954, Vol 96, 53) and not like that drawn up by Kell and Hellicar (Akustika, 1956, Vol 6, Nr 2, p 232).

There are 8 figures, 2 tables and 26 references, 10 of which are Soviet, 2 German and 14 English.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova  
(Karpov Physico-chemical Institute) and NII MRTF

SUBMITTED: July 18, 1957

Card 3/3

AUTHORS: Yezhkova, Z.I., Zhdanov, G.S. and Umanskiy, M.M. 70-3-2-18/26

TITLE: X-ray Determination of the Thermal Expansion Coefficients of  
Guanidine Aluminium Sulphate Hexahydrate -  $C(NH_2)_3[Al(H_2O)_6] \cdot [SO_4]_2$  - (GASH) (Rentgenograficheskoye opredeleniye  
koeffitsiyentov termicheskogo rasshireniya guanidin-alyuminiy-  
sul'fata geksagidrata -  $C(NH_2)_3[Al(H_2O)_6] \cdot [SO_4]_2$ )

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 2, pp 231-232 (USSR)

ABSTRACT: The lattice parameters of GASH at  $25^{\circ}C$  were found to  
be  $a=11.7159 \pm 0.0007$  KX,  $c=8.9335 \pm 0.0007$  KX and the  
coefficients of thermal expansion were determined as  
for the  $a$  axis,  $\alpha_{perp.} = 10.0 \pm 0.4 \times 10^{-6}$  per deg. C

for the  $c$  axis,  $\alpha_{par.} = 93.3 \pm 1.0 \times 10^{-6}$  per deg. C

The expansion was measured over the range  $25-55^{\circ}C$ . The 13.0.0.  
Card/l reflections were used with Cu and Fe radiations,  
and the 009 reflections were used with Cu and Fe radiations,  
respectively. There are 1 figure and 2 Soviet and 2 English refs.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova  
(Moscow State University im. M.V. Lomonosov)  
SUBMITTED: July 3, 1957.

AUTHORS: Belov, N.V., Belyayev, L.M., Bokiy, G.B., Bronnikova, Ye.G.,  
Vaynshteyn, B.K., Zhdanov, G.S., Iveranova, V.I., Kitaygorod-  
skiy, A.I. and Pinsker, Z.G. 70-3-2-26/26

TITLE: The Fourth International Congress of Crystallography  
(IV mezhunarodnyy kongress kristallografov) (Montreal,  
July 10-19, 1957)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 2, pp 250 - 260  
(USSR).

ABSTRACT: Outline of the scientific proceedings of the  
conference.

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USCOMM-DG-60577

AUTHORS: Varfolomeyeva, L.A., Zhdanov, G.S. and Umanskiy, M.M. 70-3-3-23/36

TITLE: The Determination in Principal of the Structure of the Isomorphous Group of Compounds  $[C(NH_2)_3][M(H_2O)_6]_2[EO_4]_2$

Where  $M = Al$  or  $Cr$  and  $E = S$  or  $Se$  (Printsipial'naya rasshifrovka strukturny izomorfnoy gruppy soyedineniy  $[C(NH_2)_3][M(H_2O)_6]_2[EO_4]_2$ ,  $M = Al, Cr; E = S, Se$ )

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 3, pp 368 - 371 (USSR).

ABSTRACT: From packing considerations possible positions for the various structural groups in compounds of the GASH type are suggested:  $M(H_2O)_6$  at  $(0,0,0)$ ,  $(1/3, 2/3, z_1)$  and  $(2/3, 1/3, z_1)$ ;  $C(NH_2)_3$  at  $(0,0,z_2)$ ,  $(1/3, 2/3, 0)$  and  $(2/3, 1/3, 0)$ ;  $EO_4$  at  $(1/3, 1/3, 1/4)$ ,  $(0, 2/3, 1/4)$ ,  $(2/3, 0, 1/4)$ ,  $(2/3, 2/3, 3/4)$ ,  $(0, 1/3, 3/4)$  and  $(1/3, 0, 3/4)$ . Patterson projections  $P(x,y)$  and  $P(x,z)$  were calculated from Weissenberg photographs for the compounds with  $(Al, S)$  and  $(Al, Se)$ . These largely confirm the suggested model. There are 3 figures and 3 tables and 3 References, 1 of which

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70-3-3-23/36  
The Determination in Principal of the Structure of the Isomorphous  
Group of Compounds  $[C(NH_2)_3][M(H_2O)_6]_2$ ,  $[EO_4]_2$  Where M = Al or Cr  
and E = S or Se

is Soviet and 2 English.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet imeni  
M.V. Lomonosova (Moscow State University imeni  
M.V. Lomonosov)

SUBMITTED: February 22, 1958

Card 2/2

AUTHORS: Zhdanov, G.S., Zhuravlev, N.N., Kuz'min, R.N. and  
Soklakov, A.I. 70-3-3-26/36

TITLE: The Establishment by X-ray Crystallography of a New Compound  
 $\text{Bi}_3\text{Rh}$  in the System Bi-Rh (Rentgenograficheskoye ustanov-  
leniye novogo soyedineniya  $\text{Bi}_3\text{Rh}$  v sisteme Bi-Rh)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 3, pp 373 - 374  
(USSR).

ABSTRACT:  $\text{Bi}_4\text{Rh}$  has been supposed to occur in three polymorphic  
modifications  $\alpha$ ,  $\beta$  and  $\gamma$ . Goniometric and X-ray obser-  
vations have been made on  $\beta$ - $\text{Bi}_4\text{Rh}$ . Its habit is identical  
with that of  $\text{Bi}_3\text{Ni}$  and its cell dimensions  $a=9.1$ ,  $b=4.2$ ,  
 $c=11.4$  Å are close to those of  $\text{Bi}_3\text{Ni}$  ( $a=8.875$ ,  $b=4.115$ ,  
 $c=11.477$ ). Both have the space group Pnma =  $D_{2h}^{16}$ .  $d_{obs.} =$   
 $= 10.7 \pm 0.2$  gm/cm<sup>3</sup>. gives Z nearly equal to 3 if the formula  
 $\text{Bi}_4\text{Rh}$  is used. It seems clear that the formula should be  
 $\text{Bi}_3\text{Rh}$  giving z=4 and powder photographs also confirm this  
Card1/2 resemblance to  $\text{Bi}_3\text{Ni}$ .

70-3-3-26/36

The Establishment by X-ray Crystallography of a New Compound  $\text{Bi}_3\text{Rh}$   
in the System Bi-Rh

There are 7 Soviet references.

ASSOCIATION: Moskovskiy gosudarstvenny universitet  
imени M.V. Lomonosova (Moscow State University  
imени M.V. Lomonosov)

SUBMITTED: July 11, 1957

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78-3-3-36/47

AUTHORS: Zhdanov, G. S., Zhuravlev, N. N., Kuz'min, R. N.

TITLE: An Investigation of the System Bi-Rh (Issledovaniye sistemy Bi-Rh)

PERIODICAL: Zhurnal Neorganicheskoy Khimii, 1958, Vol. 3, Nr 3, pp. 750-754  
(USSR)

ABSTRACT: By X-ray analysis some compounds of bismuth and rhodium, especially  $\text{Bi}_4\text{Rh}$  and  $\text{Bi}_2\text{Rh}$  were investigated. The crystals of  $\beta\text{-Bi}_4\text{Rh}$  and  $\alpha\text{-Bi}_2\text{Rh}$  were investigated. The  $\beta\text{-Bi}_4\text{Rh}$ -crystals are rhombic and have the following lattice constants:  $a = 11,4 \pm 0,2 \text{ \AA}$ ,  $b = 9,0 \pm 0,2 \text{ \AA}$ ,  $c = 4,2 \pm 0,1 \text{ \AA}$ . The density of the crystals is  $\rho = 10,7 \pm 0,2 \text{ g/cm}^3$ . The compound  $\alpha\text{-Bi}_2\text{Rh}$  crystallizes in the monoclinic system with periods  $a = 6,7 \text{ \AA}$ ,  $b = 6,8 \text{ \AA}$ ,  $c = 6,9 \text{ \AA}$ . The system CoSb was investigated at the same time and on the basis of X-ray analyses it was found that  $\alpha\text{-Bi}_2\text{Rh}$  and CoSb<sub>2</sub> are isomorphous compounds. The crystals  $\alpha\text{-Bi}_2\text{Rh}$  can be classified with the structural type of markasite. During the melt-

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78-3-3-36/47

An Investigation of the System Bi-Rh

ing of antimony with cobalt, rhodium and iridium crystals of  $\text{CoSb}_2$ ,  $\text{RhSb}_2$  and  $\text{IrSb}_2$  were determined. These crystals have a monoclinic structure. The unit cells of the compounds  $\text{CoSb}_2$ ,  $\text{RhSb}_2$  and  $\text{IrSb}_2$  were calculated by the roentgengoniometric method.

$\text{CoSb}_2$  :  $a = 6,5 \text{ \AA}$ ,  $b = 6,5 \text{ \AA}$ ,  $c = 6,5 \text{ \AA}$ ,  $\beta = 118 \pm 1^\circ$ ,  $\delta = 8,3 \text{ g/cm}^3$ ,  $Z = 4$

$\text{RhSb}_2$  :  $a = 6,6 \text{ \AA}$ ,  $b = 6,5 \text{ \AA}$ ,  $c = 6,7 \text{ \AA}$ ,  $\beta = 117 \pm 1^\circ$ ,  $\delta = 9,0 \text{ g/cm}^3$ ,  $Z = 4$

$\text{IrSb}_2$  :  $a = 6,6 \text{ \AA}$ ,  $b = 6,5 \text{ \AA}$ ,  $c = 6,7 \text{ \AA}$ ,  $\beta = 116 \pm 1^\circ$ ,  $\delta = 10,85 \text{ g/cm}^3$ ,  $Z = 4$

$\text{Bi}_{12}\text{Rh}$  :  $a = 6,7 \text{ \AA}$ ,  $b = 6,8 \text{ \AA}$ ,  $c = 6,9 \text{ \AA}$ ,  $\beta = 117 \pm 2^\circ$ ,  $\delta = 12 \text{ g/cm}^3$ ,  $Z = 4$

A new group of isomorphous compounds  $\text{CoSb}_2$ ,  $\alpha\text{-RhSb}_2$ ,  $\text{IrSb}_2$  and  $\alpha\text{-Bi}_{12}\text{Rh}$  was produced. There are 5 figures, 1 table, and 12 references, 11 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvenny universitet im. M. V. Lomonosova  
(Moscow State University imeni M. V. Lomonosov)

SUBMITTED: June 25, 1957

Card 2/2

ZHDANOV, G.S.

ZHURAVLEV, N.N.; ZHDANOV, G.S.

Study of the bismuth - palladium system. Zhur. neorg.khim. no.3:755  
1958. (MIRA 11:4)

(Bibliography-Bismuth-palladium alloys)

SOV/70-3-4-11/26

AUTHORS: Venevtsev, Yu.N., Zhdanov, G.S., Solov'yev, S.P. and Zubov, Yu.A.

TITLE: The Internal Fields in Certain Ferro-electrics with Structures of the Perovskite Type (Vnutrenniye polya v nekotorykh segnetoelektrikakh so strukturoy tipa perovskita)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 4, pp 473-479 (USSR)

ABSTRACT: An analysis of the methods of calculating the internal field in ferro-electrics of the perovskite type is made. The internal fields and the spontaneous polarisation in the tetragonal modifications of  $\text{BaTiO}_3$  and  $\text{PbTiO}_3$  are calculated and the influence of certain cation parameters on these quantities is estimated. The structure was assumed, as a first approximation, to be built up of point charges and point dipoles. Kozlovskiy's method (Zh.Tekh. Fiz., Vol 21, Nr 11, p 1388, 1951) where the five different ions are attached to five separate sub-lattices was used. In  $\text{BaTiO}_3$  the Ba ion was taken as the origin but in the  $\text{PbTiO}_3$  the Ti in view of the reported displacements (Shirane, Pepinsky and Danner, Acta Crystall, 1956, Vol 9, p 131). Published polarisabilities were used.

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SOV/70-3-4-11/26

The Internal Fields in Certain Ferro-electrics with Structures of  
the Perovskite Type

As the effective ionic charges were not known, a coefficient of charging  $\gamma$  ( $0 \leq \gamma \leq 1$ ), identical for all ions, was introduced. If for  $\text{BaTiO}_3$ ,  $\gamma$  was taken as 1, then the calculated, spontaneous polarisation was twice the observed value. The value  $\gamma = 1/2$  was therefore used for both  $\text{BaTiO}_3$  and  $\text{PbTiO}_3$ . The spontaneous polarisation when calculated was then near to the observed value and the internal fields were found to be  $\text{BaTiO}_3$ : Ba, 0.04; Ti, 4.84; O<sub>I</sub>, 3.66; O<sub>II</sub> and O<sub>III</sub>, 0.55.  $\text{PbTiO}_3$ : Pb, 1.83; Ti, 8.62; O<sub>I</sub> = 7.02; O<sub>II</sub> and O<sub>III</sub>, 2.23: in each case  $\times 10^8$  V/cm. As the calculations were carried out with structure coefficients  $C_{ik}$  appropriate to a cubic structure, the approximation will be much better in the case of  $\text{BaTiO}_3$  with  $c/a=1.01$  than for  $\text{PbTiO}_3$  with  $c/a = 1.06$ . The calculations were repeated with

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The Internal Fields in Certain Ferro-electrics with Structures of  
the Perovskite Type

variations in certain of the parameters. For BaTiO<sub>3</sub>,  
a was varied 4.2 and to 3.9 Å;  $\alpha$  (polarisability)  
of the Ti was doubled and halved; the charge distribution  
was tried as Ba<sup>+1/2</sup>, Ti<sup>+2.5</sup>; the polarisability  $\alpha_{Ba}$   
of the Ba ions was doubled and halved. Similar variations  
were made for PbTiO<sub>3</sub>. The relative influences of the  
various contributory effects were then apparent. The  
effects on the spontaneous polarisation were also found.  
The results are compared with those of other authors.  
There are 3 tables and 33 references, 13 of which are  
Soviet, 15 English and 5 German.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova (Institute of Physical Chemistry imeni L. Ya. Karpov)

SUBMITTED: July 18, 1971.

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SOV/70-3-6-11/25

AUTHORS: Zhdanov, G.S., Zubov, V.G., Ivanov, A.T. and Firsova, M.M.

TITLE: On the Elastic Properties of Quartz Irradiated by Neutrons  
(Ob uprugikh svoystvakh kvartsa, obluchennogo neytronami)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 6, pp 720-725 (USSR)

ABSTRACT: The elastic constants of quartz, irradiated in a reactor by fast neutrons, have been measured by the method of Bergmann and Schaeffer. After irradiation by

$2 \cdot 10^{19}$  neutrons/cm<sup>2</sup> increasing errors which lay in the limits of 0.9 to 1.7% for a relative decrease in the density of quartz of 0.18% were found in the experiment for measuring the elastic constants. Comparison with the temperature variation of the elastic constants showed that the temperature and radiation changes in the elastic constants corresponding to the same change in density were sharply distinguished. The results agree qualitatively with the work of Mayer and Gigon (J. Phys. Rad., 1957, Vol 18, p 109) on the elastic moduli of irradiated quartz. Measurements were made on blocks about 20 x 20 x 4 mm cut perpendicular to the crystallographic axes. Four series each of three plates were used, careful controls being kept. The frequencies used were 8-10 Mc/s. Wittels and Sherill (Phil. Mag., 1957, Vol 48, p 24) contrasted the

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'On the Elastic Properties of Quartz Irradiated by Neutrons

changes in the elastic constants produced by thermal and radiation-produced expansion of the crystal lattice. Although qualitatively the anisotropy is the same the actual values for it are quite different. This is shown experimentally. The structural meaning of the results obtained is not discussed. Acknowledgments to Academician I.K. Kikoin and V.L. Karpov. There are 4 tables. There are 11 references, 3 of which are Soviet, 8 English.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im.  
M.V. Lomonosova (Moscow State University imeni  
SUBMITTED: M.V. Lomonosov)  
June 12, 1958

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SOV/70-3-6-19/25

AUTHOR: Venetsev, Yu.N. and Zhdanov, G.S.

TITLE: Crystal-chemical Analysis of the Temperature Phase Transitions in Ferro- and Antiferro-electrics with Structures of the Perovskite Type (Kristallokhimicheskiy analiz temperaturnykh fazovykh perekhodov v segneto-i antisegnetoelektrikakh so strukturoy tipa perovskita)

PERIODICAL: Kristallografiya, 1958, Vol 3, Nr 6, pp 751-752 (USSR)

ABSTRACT: It is known that the phase transitions undergone by  $\text{BaTiO}_3$  and  $\text{PbTiO}_3$  on cooling follow different sequences (cubic-tetragonal-monoclinic-pseudorhombohedral and cubic-rhombohedral-monoclinic, respectively). Perovskites can be crystallo-chemically characterised by the values of the tolerance factors  $t$  relating to their packings. In  $\text{BaTiO}_3$ ,  $t$  is greater than 1 and Ti is the ferro-electric ion. In  $\text{PbZrO}_3$ ,  $t$  is less than 1 and the Pb ion is the ferro-electric one. The co-ordinations and situations of the two types of ion are quite different and therefore so are the displacements which the structure can undergo on cooling. Earlier observations on this point by the present authors is recalled. In it the sequence

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Crystal-chemical Analysis of the Temperature Phase Transitions in  
Ferro- and Antiferro-electrics with Structures of the Perovskite  
Type

of deformations was discussed as a function of  $t$   
(Kristallografiya, 1957, Vol 2, p 233).

There 11 references, 7 of which are Soviet, 3 English  
and 1 French.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova  
(L.Ya. Karpov Physico-chemical Institute)

SUBMITTED: July 18, 1957

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SOV/78-3-11-19/23

AUTHORS: Epel'baum, V. A., Sevast'yanov, N. G., Gurevich, M. A.,  
Ormont, B. F., Zhdanov, G. S.

TITLE: II. On the Phases Formed in the System Chromium-Boron (II. O  
fazakh, obrazuyushchikhsya v sisteme khrom-bor)

PERIODICAL: Zhurnal neorganicheskoy khimii, 1958, Vol 3, Nr 11, pp 2545-2552  
(USSR)

ABSTRACT: The compounds formed in the system chromium-boron are investigated. The investigations were carried out by means of chemical, radiographic, and metallographic methods in the region of the phase diagram of the system chromium-boron and in the range  $\text{CrB}_{0,35}-\text{CrB}_3$ . Purest boron (99,6%) produced by the thermal dissociation of diboranes served as initial components for the production of the chromium-boron phases. The results of the chemical and radiographic analyses of the samples were obtained by heating at 1150°C in vacuum and then at 1300°C in an argon atmosphere for 36 hours. The results are given in table 2. It was found that the  $\gamma$ -phase occurs with a rhombic lattice in the sample with a boron content of  $\text{CrB}_{0,35}-\text{CrB}_{0,58}$ . In the samples

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II. On the Phases Formed in the System Chromium-Boron

with a boron content of  $\text{CrB}_{0,41}$ - $\text{CrB}_{0,51}$  only the  $\gamma$ -phase exists. In the samples with a boron content of  $\text{CrB}_{0,55}$ - $\text{CrB}_{1,05}$  the  $\delta$ -phase ( $\text{Cr}_5\text{B}_3$ -phase) is formed. In the samples with a boron content of  $\text{CrB}_{0,59}$ - $\text{CrB}_{0,63}$  only the  $\delta$ -phase is formed. In the samples with a boron content of  $\text{CrB}_{0,68}$ - $\text{CrB}_{1,50}$  the  $\epsilon$ -phase occurs (CrB with rhombic lattice). In the samples of the composition  $\text{CrB}_{0,96}$ - $\text{CrB}_{1,13}$  no other phases were found besides the  $\epsilon$ -phase. In the sample with a boron content of  $\text{CrB}_{1,20}$ - $\text{CrB}_{1,90}$  a  $\zeta$ -phase with rhombic lattice is formed. In the sample of the composition  $\text{CrB}_{1,50}$ - $\text{CrB}_{1,65}$  no other phases were found to exist besides the  $\zeta$ -phase. In the samples with  $\text{CrB}_{1,70}$  and  $\text{CrB}_{1,90}$  only the  $\eta$ -phase is formed.

There are 2 figures, 5 tables, and 27 references, 1 of which is Soviet.

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24(2), 24(3)

AUTHORS: Venevtsev, Yu. N., Zhdanov, G. S.,  
Solov'yev, S. P.

SOV/48-22-12-17/33

TITLE:

Effect of Various Factors Upon the Curie Temperature of  
Piezoelectrics With the Structure of the Perovskite Type  
(Vliyaniye razlichnykh faktorov na temperaturu Kyuri  
segnetoelektrikov so strukturoy tipa perovskita)

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1958,  
Vol 22, Nr 12, pp 1476-1482 (USSR)

ABSTRACT:

The question concerning the Curie (Kyuri)  $T_k$  temperature of piezoelectrics having a perovskite structure, was already investigated earlier (Refs 1-7 and 8-11). The analysis of these papers shows that there is no uniform opinion on this problem. The conclusions drawn from references 1-7 are based on the assumption that in the investigated piezo- and anti-piezoelectrics the cations of the B-type are piezoactive. Actually, in some piezoelectrics the B-cations, and in others the A-cations, are piezoactive (Refs 8, 9, 13). As it was already stated (Ref 10), the results given in references 1-7 must be subjected to further examinations, because of the reason mentioned above. The classification of the piezo- and anti-piezoelectrics with perovskite structure depending on the

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Effect of Various Factors Upon the Curie Temperature  
of Piezoelectrics With the Structure of the Perovskite Type

SOV/48-22-12-17/33

type of the piezoactive cation , as proposed in references 8, 9, 13 promotes the solution of the problem discussed. Therefore, they have been investigated again in this paper. It was attempted to explain the differences of the Curie temperature in piezoelectrics with perovskite structure in the same way, by using only such characteristics as polarizability, charges and radii. The degree of covalence of the bindings in the respective compounds was neglected. The authors started from an approximate theoretical estimation. They used the results from references 15, 16, determined in the investigation of the effect of various parameters of cations upon the internal field of piezoelectrics with perovskite structure. The conclusions drawn on the basis of theoretical estimations agree with the few experimental results obtained by the authors of this article and Sawaguchi (Ref 7). As soon as new experimental data will be obtained it will be possible to check also the theoretical results still improved.

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Effect of Various Factors Upon the Curie Temperature  
of Piezoelectrics With the Structure of the Perovskite Type SOV/48-22-12-17/33

There are 2 tables and 26 references, 12 of which are Soviet.

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova  
(Physico-Chemical Institute imeni L. Ya. Karpov)

Card 3/3

VIASHEKO, V.I.; ZHDANOV, G.S.

Using calculating machines for radiographic studies. Zav. lab. 24  
no.5:634-636 '58.  
(MIRA 11:6)

1. Fiziko-khimicheskiy institut im. L.Ya. Karpova.  
(Radiography) (Electronic calculating machines)

AUTHORS:

Zhdanov, G. S., Kefeli, L. M.

76-32-3-26/43

TITLE:

An Investigation of the Texture of Copper Obtained in the Leaching of CuAl<sub>2</sub> by the Method of Pole Figures  
(Issledovaniye tekstury medi pri vyschelachivaniyu CuAl<sub>2</sub> metodom polyusnykh figur)

PERIODICAL:

Zhurnal Fizicheskoy Khimii, 1958, Vol 32, Nr 3,  
pp 666 - 669 (USSR)

ABSTRACT:

It was noticed that copper in a polycrystalline, dispersed form remains behind in the leaching out of aluminium from alloys. Radiographic investigations showed condensed spots which indicate a certain orientation of the copper crystals. The method mentioned in the title was employed for the investigation of the crystal orientation, because it is convenient and objective. A schematically drawn representation of the arrangement in the method of investigation according to the pole figures is given. From the data of the method of investigation, follows that the crystal sample was leached for 1-10 minutes with a 20% lye at 40-50°C. For obtaining

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76-32-3-26/43

An Investigation of the Texture of Copper Obtained in the Leaching of  
 $\text{CuAl}_2$  by the method of Pole Figures

the polar points, a series of X-ray photographs was taken, where the sample was turned in a circle by  $10^{\circ}$  (around the 001 axis). From the considerations it follows that in leaching out the monocrystal  $\text{CuAl}_2$ , on the one hand, the formation of polycrystalline  $\text{CuAl}_2$  takes place, and on the other hand that of copper, where the two crystals form a texture. The texture of copper is characterized by the following data:  $\text{CuAl}_2 /001/, /100/, /010/$ ; Cu  $/011/, \sim /111/, \sim /111/$ . There are 9 figures, 3 tables, and 6 references, 3 of which are Soviet.

SUBMITTED: December 11, 1956

Card 2/2

56-34-4-5/6o

AUTHORS: Zhuravlev, N. N., Mingazin, T. A., Zhdanov, G. S.

TITLE: The Structure of Superconductors. XII (Struktura sverkhprovodnikov. XII) The Investigation of Bismuth - Rubidium Alloys (Issledovaniye splavov vismuta s rubidiem)

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1958,  
Vol. 34, Nr 4, pp. 820 - 826 (USSR)

ABSTRACT: According to thermal, microscopical, and radiographic data the test diagram (probnaya diagramma) of the fusibility of the system Bi-Rb is constructed. The investigation of this system is connected with the solution of various methodical problems. The main difficulties mainly are connected with the high chemical activity of metallic rubidium and also with the great difference of the physical-chemical properties of bismuth and rubidium. The whole investigation was performed at small quantities ( $\sim 3g$ ) of rubidium which required the working out of micromethods for the production of the alloys and their physical-chemical analysis. First the production

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The Structure of Superconductors. XII. The Investigation of Bismuth-Rubidium Alloys

56-34-4-5/60

of the alloys is discussed, the apparatus used for this are discussed by means of figures. The melting was performed in small resistance furnaces. Then the authors report on the thermal analysis of the alloys. The curves of heating and cooling were taken by an automatic electron potentiometer unto a temperature of 50 - 100°C. The results of the thermal analysis obtained are illustrated in a diagram and subsequently discussed. At the bismuth-rich alloys (to the composition Bi<sub>2</sub>Rb) also a metallographical investigation was performed. According to this the number of the crystals of the compound Bi<sub>2</sub>Rb increases with increasing rubidium content in the alloy. The alloy with 15,8 per cent by weight rubidium according to its composition resembles the compound Bi<sub>2</sub>Rb (17 per cent by weight;) it contains a small quantity of eutectic and is almost homogeneous. In the system Bi-Rb the various phases differ by their color. Then the authors report on the determination of the structure of the superconductive compound Bi<sub>2</sub>Rb. By

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The Structure of Superconductors. XIII. The Investigation of Bismuth-Rubidium Alloys

56-34-4-5/60

exact determination of the lattice period of  $\text{Bi}_2\text{Rb}$  the value  $a = 9,590 \pm 0,002 \text{ kX}$  was obtained. The distances between the lattice planes, computed from these data agree well with the measured results. The Rb atoms in the structure of the  $\text{Bi}_2\text{Rb}$  are distributed according to the diamond law. Some conclusions are: In the system bismuth-rubidium 4 compounds were stated:  $\text{Bi}_2\text{Rb}$ ,  $\text{BiRb}_3$ , and two compounds of the probable composition  $\text{Bi}_2\text{Rb}_3$  and  $\text{BiRb}_2$ . The maxima in the fusibility diagram correspond to the compounds  $\text{Bi}_2\text{Rb}$  and  $\text{BiRb}_3$ . These compounds form in the fusion of the components with high heat emission. The two other compounds form according to the peritectic reaction. The superconductive compound  $\text{Bi}_2\text{Rb}$  crystallizes in isometric syngony with  $a = 9,590 \pm 0,002 \text{ kX}$  and has a structure of the type of  $\text{Cu}_2\text{Mg}$ . The increase of the minimum interatomic distances in the  $\text{Bi}_2\text{Rb}$  leads to an increase of the temperature of the transition into the superconductive state. Finally the author

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The Structure of Superconductors. XII. The Investigation of Bismuth-Rubidium Alloys 56-34-4-5/60

thanks Professor N. Ye. Alekseyevskiy for his valuable advice in the performance of this work, and R. N. Kuz'min for his assistance in the performance of the experiments.- There are 6 figures and 10 references, 9 of which are Soviet.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet (Moscow State University)

SUBMITTED: November 15, 1957

1. Bismuth alloys--Analysis 2. Superconductors--Structural analysis

Card 4/4

AUTHORS: Gol'der, G. A., Zhdanov, G. S. 20-118-6-23/43

TITLE: A Radiographic Structural Examination of Naphthazarine  
(Rentgenostrukturnoye issledovaniye naftazarina)

PERIODICAL: Doklady Akademii Nauk SSSR, 1958, Vol. 118, Nr 6,  
pp. 1131-1133 (USSR).

ABSTRACT: Naphthazarine is widely spread amongst the hydroxy ketone dyes, the naphthoquinone-derivates (5,8-dioxy-1,4 naphthaquinone). The authors give a survey on the literature of the radiographic investigation (reference 1), as well as of the investigation of magnetic anisotropy (reference 2). There exist 2 modifications of naphthazarine: 1) dark-green pins, 2) dark-red prisms; the authors obtained a third modification by crystallization in benzene, viz. 3) light-red platelets. Their radiographic data are given in table 1. If and when all 3 modifications precipitate from the solution, well-faceted crystals of the 3rd modification occur most frequently. The modifications 1) and 2) were formed by sublimation. The belonging of the crystals of the 1st modification to the spatial group  $C_{2h}^6 = P\bar{2}_1/c$  and the presence of 2 molecules

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A Radiographic Structural Examination of  
Naphthazarine

20-118-6-23/43

in the elementary mesh conform the assumption (reference 1) that a center of symmetry exists in the molecule of the crystals of the 1st modification. The introduction of an inner hydrogen compound O . . . H-O in the conjugated bond-system must have caused an essential change of the  $\pi$ -electronic interaction in the whole molecule. This must, in return, lead to a redistribution of the electronic density in the molecule. A complete radiographic analysis of the crystals of this modification was interesting therefore. The lengths of the bonds between the atoms in the molecule were computed (II) from the atomic coordinates computed from  $\rho$  (OKI)(table 2). The computations of the distances between the atoms were carried out under the assumption that the molecule of the surface  $yz$  lies parallel. The angle formed by the bond-line  $C_9 - C_{10}$  with the y-axis of the mesh, is  $50^\circ$ . The smallest distance between the carbon- and oxygen-atoms in various molecules is 3,10 Å. The results of the radiographic structural analysis confirm the presence of a center of symmetry in the 1st modification of naphthazarine. As mentioned above, all 3 modifications precipitate simultaneously with the crystallization of the solution: 2 centro-symmetrical ones (A), and a none-centro-symmetrical one (B). The

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A Radiographic Structural Examination of Naphthazarine 20-118-6-23/43

recrystallization of each of these modifications leads in return to the formation of all these 3 modifications, though one of them prevails largely. It may thus be presumed that the transition of an isomer of an A-structure into an isomer of a B-structure (and viceversa) takes place. This transition is explained with scheme III and was presumed in reference 4. The orientation in space of the molecule in the yz-surface achieved by the authors, is very similar to that for the centro-symmetrical modification 2) given in reference 3. A three-dimensional synthesis is required for determining the 3rd coordinate x and for defining precisely the obtained results.  
There are 1 figure, 2 tables, and 4 references, 1 of which is Slavic.

ASSOCIATION: Physico-Chemical Institute imeni L. Ya. Karpov  
(Fiziko-khimicheskiy institut im. L. Ya. Karpova)

PRESENTED: November 20, 1957, by N. V. Belov, Academician.

SUBMITTED: August 16, 1957.  
Card 3/4

"APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R002064620006-5

A Radiographic Structural Examination of  
Naphthazarine

20-118-6-23/43

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APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R002064620006-5"

20-119-1-23/52

AUTHORS: Dokunikhin, N. S., Gol'der, G. A., Zhdanov, G. S.

TITLE: The Radiographic Investigation of 1,4-di-Anilido-Anthraquinone and 1,4-Dimesido-Anthraquinone (Rentgenograficheskoye issledovaniye 1,4-dianilidoanthrakhinona i 1,4-dimesidoanthrakhinona)

PERIODICAL: Doklady Akademii Nauk SSSR, 1958, Vol. 119, Nr 1,  
pp. 87 - 89 (USSR)

ABSTRACT: Sulfo acids of 1,4-di-(arylamino)-anthraquinone form an important group of solid dyes for wool. The majority of the 1,4-di-(arylamino)-substitutes of anthraquinone are green. An exception is made by the derivatives in which all hydrogen atoms, in an ortho-position, of the aryl-residues are substituted. Such compounds as well as the corresponding alkyl-amino-and hydro-aryl-amino-derivatives have an intensive bright-blue color. In the presence of methyl-ethyl-groups or of bromine atoms in all ortho-positions of the phenyl residues or in the position of 2,3-anthraquinone respectively

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20-119-1-23/52

The Radiographic Investigation of 1,4-di-Anilido-Anthraquinone and  
1,4-Dimesido-Anthraquinone

the absorption in the short-wave range of light is absent and the chief maximum is displaced in the direction of the short waves, when the light absorption is measured by  $\text{CCl}_4$ -solutions of 1,4-di-(arylamino)-anthraquinone in the visible and in the ultraviolet range of the spectrum (Reference 1). The appearance of an additional principal band and the deepening of the principal band in the absence of spatial disturbances might logically be considered a consequence of the coplanarity of the molecule. This is also indicated by the comparison of the absorption frequencies in the infrared spectral region (Reference 2). These data indicate the weakening of the inner-molecular hydrogen bond of the carbonyl-oxygen with the hydrogen of the amino groups in the presence of spatial obstacles of a coplanar orientation of the benzene nuclei. This bond is weakened by the increased distance due to the leaving of the plane of the anthraquinone-

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20-119-1-23/52

The Radiographic Investigation of 1,4-di-Anilido-Anthraquinone and  
1,4-Dimesido-Anthraquinone

cycles by hydrogen and is caused by the disturbance of the conjugation -system (Reference 3). It would be desirable to find a direct proof of the flat structure of the molecules of 1,4-di-(aryl amino)-anthraquinone in the absence of spatial difficulties. For the purpose of deciding the problem of coplanarity of the benzene nuclei with the plane of the basic part of the molecule, crystals of both compounds mentioned in the title were radiographically measured. The results are given in table 1. From the dimensions of the elementary cell of the first compound can be assumed that the basic part of the molecule is here entirely or almost parallel with the ac-plane, as axis b is the shortest one (8,73 Å). From the conditions of symmetry of the spatial group

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$$C_{2h}^5 = P2_1/c$$

20-119-1-23/52

The Radiographic Investigation of 1,4-di-Anilido-Anthraquinone and  
1,4-Dimesido-Anthraquinone

follows that a slip plane with a displacement along axis c runs vertical to axis b. Thereby the 4 molecules occurring in the unit cell are orientated in layers which are perpendicular to axis b. A variant of this orientation is shown by figure 1. It admits a slight turn of the benzene nucleus in relation to the other part of the molecule as well as a certain possible turn of the entire molecule in relation to the plane ac. Thus the packing of the molecules in the crystal does not require an additional change of the angle of rotation of the benzene nucleus as compared to the free molecule. The shortest axis in the crystal of the second compound is the a-axis (7,98 Å). Its length corresponds to the dimensions of the benzene nucleus and to the  $\text{CH}_3$ -groups connected with it (8,8 Å). A solid packing of molecules in the crystal and the fulfillment of the conditions of symmetry of the spatial group for molecules of the second compound

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20-119-1-23/52

The Radiographic Investigation of 1,4-di-Anilido-Anthraquinone and  
1,4-Dimesido-Anthraquinone

of the methyl groups to all meta-positions of the benzene nucleus creates so great spatial difficulties that the coming out with the anthraquinone cycles from the coplanarity amounts to almost 90°. Thereby the inner-molecular linkage is considerably weakened. There are 2 figures, 1 table, and 5 references, all of which are Soviet.

ASSOCIATION: Nauchno-issledovatel'skiy institut organicheskikh poluproduktov i krasiteley im. K. Ye. Voroshilova (Scientific Research Institute of Organic Semiproducts and Dyes imeni K. Ye. Voroshilova). Nauchno-issledovatel'skiy fiziko-khimicheskiy institut im. L. Ya. Karpova (Scientific Physical-Chemical Research Institute imeni L. Ya. Karpov)

PRESENTED: November 20, 1957, by N. V. Belov, Member, Academy of Sciences, USSR

SUBMITTED: August 16, 1957

Card 6/6

ZHDANOV, G.S.

Moscow. Vysshaya tekhnicheskaya uchilishche imeni Baumana. Izdatsa

Naukno-tekhnicheskikh mashin  
Vyschitatel'naya tekhnika (Computer Techniques) Moscow, Maschis, 1959.  
153 p. (Series: Moscow. Vysshaya tekhnicheskaya uchilishche.  
Sbornik. No. 2) 2,500 copies printed.

Ed.: B.V. Andrianov, Candidate of Technical Sciences; Tech. Eds.:  
D.I. Model' and A.P. Ovrovov; Managing Ed. for Literature on  
Machine Building and Instrument Construction: M.V. Polozovskiy.

PURPOSE: This book may be useful to aspirants and other students  
specializing in computer technology, and also to designers and  
engineers and technical personnel who make use of electronic  
computers.

School Iosif Baumana) In honor of the 40th anniversary of the  
October Revolution. The articles contain the results of character-  
istic and experimental studies on the performance of various com-  
ponents of electronic computers. Among the topics discussed are  
program storage, control devices, the connection between the parts  
of hardware of an algorithm and a machine, etc. The application of  
these components to the control of technological processes is  
also discussed.

Author: V.N. Gulybin, I.V. Gavrilov, N.M. Golubin,  
Candidate of Technical Sciences; Analysis of The Quality of Service  
Systems With Discrete Element

Dobrov, Ye.Ya. Engineer. The Effect of Block Diagram Parameters on  
the Performance Quality of a Tubular Circuit Operational  
Amplifier

Andrianov, B.V. Candidate of Technical Sciences; V.I. Golubin,  
Candidate of Technical Sciences, and Yu.K. Dovzhenko, Engineer.  
Device for Realizing the Form of Recording of Programs  
Treblikov, M. Candidate of Technical Sciences, and A.G. Serebryakov,  
Engineer. Certain Principles of Constructing Local Control by  
External Memory Devices

Vlasenko, V.I., Candidate of Technical Sciences, O.I. Gulybin,  
Professor, A.M. Danilev, Engineer, and L.M. Litsyn, Engineer.  
Method of Forming the Images of Numbers by Means of a Perforated  
Matrix

Shestopalov, Vl. Candidate of Physical and Mathematical Sciences.  
The Connection Between the Parameters of an Algorithm and of a  
Machine

Andrianov, B.V. Candidate of Technical Sciences; V.I. Golubin,  
Candidate of Technical Sciences, and A.P. Savel'yan, Engineer.  
Device for the Control of Recording of Information on Magnetic Tape

Vasil'yev, O.V. Engineer. Analysis of Certain Relationships for  
an Economic Selection of the Dimensions of a Magnetic Drum

Andrianov, B.V. Candidate of Technical Sciences, and Yu.V.  
Vlasenko, Engineer. On the Problem of the Exactness of the Re-  
presentation of Continuously Varying Values in a Numerical Code

Shestopalov, Yu. A. Candidate of Physical and Mathematical Sciences.  
Solution of Boundary Value Problems by the Method of Polynomial  
Approximations

Jarbas, O.I. Engineer. Certain Considerations on the Preventive  
Control of Electronic Computers

K.J. Soplun, Engineer. Photoelectric Device Which Receives  
Printed Numerical Signs

Palashovskiy, A.M. Engineer. Analysis of Information Stores  
Components of Computers

Chevernikov, V.N. Candidate of Technical Sciences. Relay  
Integrating Device With Electromagnetic Power Clutch

Kalemenkov, V.A. Engineer. Certain Algorithms for the Rational  
Planning of Production

Dmitriev, M.M. Candidate of Technical Sciences. Circuit

Mechanisms for Programmed Control

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ZHURAVLEV, D. B.; SOLOV'EV, Sergey Pavlovich; VINOGRADOV, Pavly Bimalayevich;  
VINOGRADOV, V. V.

"Internal Fields in the Orthorhombic Modification of  
Barium Titanate"

A Report presented at Symposium of the International Union of  
Crystallography Leningrad, 22-27 May 1979

VLAZENKO, V.I., kand.tekhn.nauk; ZHDANOV, G.S., prof.; DEMENT'YEV,  
A.M., inzh.; ANTONOVA, I.M., inzh.

Use of a ferrite matrix in a method for forming numbers.  
[Trudy] MVTU no.2:64-69 '59. (MIRA 13:5)  
(Electronic calculating machines)

24,7700  
24 (3), 10 (6)

68047

SOV/55-59-3-15/32

AUTHORS: Alekseyevskiy, N. Ye., Zhdanov, G. S., Zhuravlev, N. N.TITLE: The Problem of the Superconductivity of Compounds of Bismuth With Alkali MetalsPERIODICAL: Vestnik Moskovskogo universiteta. Seriya matematiki, mehaniki, astronomii, fiziki, khimii, 1959, Nr 3, pp 113 - 115 (USSR)

ABSTRACT: By analysis of the critical temperature of bismuth-alkali compounds a linear dependence between the critical temperature  $T_c$  of the compound and the atomic radius of the alkali metal was found. This linear dependence does not hold for the LiBi compound; it possibly holds for the compounds with alkaline earth metals. A comparison between  $T_c$  and the atomic radius is quite permitted for the isomorphic compounds  $KBi_2$ ,  $RbBi_2$ ,  $CsBi_2$ , but less for  $NaBi$ , which has a different type of lattice. With an increase in the critical temperature  $T_c$  the minimum interatomic spacings in the transition from  $KBi_2$  to  $CsBi_2$  increase, the dependence between the minimum distance and  $T_c$  being linear. By means of a relationship between  $dH_k/dT$  and  $\gamma$  (the coefficients

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24.7700

5.4130

5 (4), 24 (3)

68048

SOV/55-59-3-16/32

## AUTHORS:

Zhuravlev, N. N., Zhdanov, G. S., Alekseyevskiy, N. Ye.

## TITLE:

The Crystal Chemistry of Superconductive Bismuth Compounds

## PERIODICAL:

Vestnik Moskovskogo universiteta. Seriya matematiki, mehaniki, astronomii, fiziki, khimii, 1959, Nr 3, pp 117 - 127 (USSR)

## ABSTRACT:

The present paper deals with some results obtained by an analysis of the crystallo-chemical data found by the authors concerning the structure of superconductive bismuth compounds. A comparison between the superconductive properties with crystallo-chemical data was carried out on the basis of known results determined in most cases by N. Ye. Alekseyevskiy (Refs 20 - 35) at the Institut fizicheskikh problem AN SSSR (Institute for Physical Problems of the AS USSR). The first part of the present paper deals with 1) the capability of bismuth to form compounds with various elements of the periodic system and 2) with the electric properties of these compounds. Among other things the existence of the compounds  $K_9Bi_7$ ,  $K_3Bi_2$ ,  $Rb_3Bi_2$ ,  $Rb_2Bi$ , and  $Cs_2Bi$  is supposed. It may be that the compounds  $CeBi_2IrBi$  and  $\beta$ -LiBi are superconductive. With the

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The Crystal Chemistry of Superconductive Bismuth Compounds SOV/55-59-3-16/32

of  $T_c$ . Bismuth compounds with transition metals show the reverse effect. The minimum interatomic spacings bismuth-bismuth vary within a wide range in the case of the various bismuth compounds. Bismuth compounds with small and also with large bismuth-bismuth spacings are not superconductive, but all compounds in which bismuth-bismuth spacings are within the "optimum" interval (3.1 to 3.8 Å) are superconductive. The superconductive modification of bismuth probably has a face-centered elementary cell. A large table shows the atomic-crystalline structural properties of bismuth compounds with non-superconductive elements. The experiments made by N. B. Brandt (Ref 56) are indicative of a decrease of the anisotropy of the crystal structure of ordinary bismuth in the case of compression. There are 3 figures, 4 tables, and 56 references, 40 of which are Soviet.

ASSOCIATION: Kafedra fiziki tverdogo tela (Chair for Solid-state Physics)

SUBMITTED: February 27, 1959

Card 3/3

4

AUTHORS: Yezhkova, Z.I., Zhdanov, G.S. and Umanskiy, M.M. SOV/70-4-2-24/36

TITLE: The Thermal Expansion of Crystals of Triglycinesulphate  
in the Region of Their Ferro-electric Transition  
(Termicheskoye rasshireniye kristalla triglycinesul'fata  
v oblasti segnetoelektricheskogo perekhoda)

PERIODICAL: Kristallografiya, 1959, Vol 4, Nr 2, pp 249-253 (USSR)

ABSTRACT:  $(\text{CH}_2\text{NH}_2\text{COOH})_3\text{H}_2\text{SO}_4$  is monoclinic with  $a = 9.15$ ,  
 $b = 12.69$ ,  $c = 5.73 \pm 0.03 \text{ \AA}$ ,  $\beta = 105^{\circ}40' \pm 20'$  with  
space group  $P2_1$  below the Curie point at  $47^\circ$  and  
 $P2_1/m$  above.  $Z = 2$   $d_{\text{obs}} = 1.69$  and the ferro-electric  
axis is [010] (according to Wood and Holden - Ref 6).  
Here, the thermal expansion of single crystals has been  
measured from X-ray single-crystal oscillation photographs.  
The most accurate values were obtained from  $d_{900}(\text{FeK}\alpha)$   
with  $\theta \sim 81^\circ$ ,  $d_{007}(\text{Cu K}\alpha)$  with  $\theta \sim 78^\circ$ ,  
 $d_{505}(\text{Ni K}\alpha)$  with  $\theta \sim 81^\circ$  and  $d_{0,14,0}(\text{Co K}\alpha)$  with  
 $\theta \sim 83^\circ$ . Absorption corrections (for the 0.4 mm dia

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SOV70-4-2-24/36

The Thermal Expansion of Crystals of Triglycinesulphate in the Region  
of Their Ferro-electric Transition

crystal were not applied. The accuracy was estimated at  $\pm 0.0015$  kX.  $\beta$  was calculated. The components of the thermal expansion tensor (principal components) were calculated as follows, where  $\varphi$  is the angle of  $\alpha_{11}$  to the a axis of the crystal. At  $25^\circ\text{C}$   $\alpha_{11} = -37$ ,  $\alpha_{22} = 5$ ,  $\alpha_{33} = 142$  (in each case  $\text{deg}^{-1} \times 10^{-6}$ ) and  $\varphi = 22^\circ$ . At  $42.5^\circ$   $\alpha_{11} = -20$ ,  $\alpha_{22} = 36.5$ ,  $\alpha_{33} = 119$  and  $\varphi = 7^\circ 20'$ . Between  $51$  and  $75^\circ$   $\alpha_{11} = 40$ ,  $\alpha_{22} = 64$ ,  $\alpha_{33} = -12.5$  and  $\varphi = 5^\circ 40'$ . The cell volume changes smoothly over the whole temperature range. It is concluded that the phase transition is of the second order and that a marked redistribution of the hydrogen bonds parallel to the ac plane occurs at the ferro-electric transition. There are 4 figures, 3 tables and 7 references, 2 of which are Soviet, 4 English and 1 international.

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SOV/70-4-2-24/36  
The Thermal Expansion of Crystals of Triglycinesulphate in the  
Region of Their Ferro-electric Transition

ASSOCIATION: Moskovskiy gosudarstvennyy universitet imeni  
M.V. Lomonosova (Moscow State University imeni  
M.V. Lomonosov)

SUBMITTED: October 13, 1958

Card 3/3

AUTHORS: Venevtsev, Yu.N., Zhdanov, G.S., SOV/70..4-2-26/36  
Ivanova, V.V. Solov'yev, S.P. and

TITLE: On Internal Fields in Ferroelectric  $PbTiO_3$  (O vnutrennikh  
polyakh v segnetoelektriike  $PbTiO_3$ )

PERIODICAL: Kristallografiya, 1959, Vol 4, Nr 2, pp 255-257 (USSR)

ABSTRACT: Calculations of the internal fields in  $PbTiO_3$  crystals  
have been made by the Madelung-Hagedorn method  
(R. Hagedorn - Ref 3) which is more accurate than  
Kozlovskiy's method used before, according to the work  
of Yu.N. Venevtsev et al (Ref 1). These fields  $E_i$  are  
 $Pb$  1.4,  $Ti$  6.9,  $O_I$  6.1 and  $O_{II}$ ,  $O_{III}$   $1.8 \times 10^8$  V/cm.  
The contributions of the separate ions to the spontaneous  
polarisation of  $81 \times 10^{-6}$  coulomb/cm<sup>2</sup> are tabulated. The  
internal fields for model crystals of the  $PbTiO_3$  type  
but with ions of different polarisability are similarly  
calculated. For  $BaTiO_3$  the calculations by both methods

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are in good quantitative agreement. Graphical examination of the parameters affecting the internal fields show their relative importance. In order they are: 1) lattice period; 2) charge on the ferroelectric cation; 3) polarisability of the ions of the oxygen octahedra; 4) polarisability of the ferroelectric cation; 5) polarisability of the non-ferroelectric cation. There are 1 figure, 2 tables and 5 references, 4 of which are Soviet and 1 German.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova  
(Physical-Chemical Institute im. L. Ya. Karpov)

SUBMITTED: November 14, 1958

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AUTHORS: Venevtsev, Yu.N., Solov'yev, S.P. and <sup>SOV/70-4-4-17/34</sup> Zhdanov, G.S.

TITLE: On the Structural Coefficients of the Internal Field in  
Ferroelectrics of the Perovskite Type

PERIODICAL: Kristallografiya, 1959, Vol 4, Nr 4, pp 575-578 (USSR)

ABSTRACT: The notation is carried over from an article by the same authors (Ref 1). The values of the projection of the structural coefficients  $S_1(x,y,z) = C_{ik}$  for different orientations (cube edges, face or body diagonals) of the dipoles in a cubic perovskite-type cell are tabulated in terms of  $P$  and  $Q$ . ( $P = -15.04102$  and  $Q = 4.33387$ ). The coefficients  $C_{ik}$  are dimensionless and numerically equal to the field strength, additional to the Lorentz field, due to the sub-lattice of unit dipoles of the  $k$ -th sort of ion and acting on the  $i$ -th sort of ion. The derivation of expressions such as :

$$S_2(0, 1/2, 1/2) = S_3(0, 1/2, 1/2) = \alpha/\sqrt{3}$$

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$S_2(1/2, 0, 0) = S_3(1/2, 0, 0) = P/\sqrt{3}$   
is given but all the other values are tabulated.

There are 1 table and 1 Soviet reference.

ASSOCIATION: Fiziko-khimicheskiy institut im. L.Ya. Karpova  
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SUBMITTED: June 23, 1958

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SOV/70-4-5-16/36

AUTHORS: Yezhkova, Z. I., Zhdanov, G. S., Umanskiy, M. M.

TITLE: An X-Ray Diffraction Method for the Determination of the Thermal Expansion Tensors of the Crystals of Low Symmetry

PERIODICAL: Kristallografiya, 1959, Vol 4, Nr 5, pp 723-726  
(USSR)

ABSTRACT: If the principal expansion directions, that determine the diagonal tensor  $\alpha_{11}$ , are the orthogonal coordinate axes  $X'$ ,  $Y'$ ,  $Z'$ , the thermal-expansion coefficient in 1 direction is described by

$$\Delta_1 = \alpha_{11}c_{11}^2 + \alpha_{22}c_{21}^2 + \alpha_{33}c_{31}^2 \quad (3)$$

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where  $c_{11}$ ,  $c_{21}$ ,  $c_{31}$  are direction cosines of 1. In cubic, tetragonal, hexagonal, rhombohedral and orthorhombic crystals, the expansion coefficients along one, two, or three crystallographic axes suffice to determine the tensor. In monoclinic crystals only

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$\langle 010 \rangle$  coincides with one of the expansion tensor components, and in triclinic crystals, none. Consequently, the determination of the thermal expansion tensor in monoclinic crystals requires the data on the expansion of interplanar spacings of 4 different  $hkl$ , more conveniently of  $d_{010}$  and of three spacings of  $d_{hkl}$  type, and in triclinic crystals of 6 different spacings, more conveniently of  $d_{100}$ ,  $d_{010}$ ,  $d_{001}$ ,  $d_{110}$ ,  $d_{101}$ ,  $d_{011}$ . In a coordinate system X,Y,Z of which the first two are parallel to  $\langle 100 \rangle$  and  $\langle 010 \rangle$ , respectively, and the third is normal to  $(010)$ , the thermal expansion of a monoclinic crystal in the direction parallel to  $(010)$  and under angle  $\varphi$  to the X axis is described by

$$\Delta_\varphi = a_{11} \cos^2 \varphi + a_{33} \sin^2 \varphi + 2a_{13} \cos \varphi \cdot \sin \varphi. \quad (2a)$$

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Here, the values are determined by the following three

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equations:

$$2\alpha_{11} = a_{11} + a_{33} + \frac{a_1 - a_3}{\cos 2\psi}; \quad 2\alpha_{33} = a_{11} + a_{33} - \frac{a_1 - a_3}{\cos 2\psi}$$

$$\tan 2\varphi = 2\alpha_{13} : (a_{11} - a_{33})$$

where  $\psi$  is the angle between  $a_{11}$  and X-axis. The orthogonal coordinate axes X,Y,Z for triclinic crystals must be chosen as follows: X coincides with the reciprocal-lattice axis  $a^*$ , Y is in the  $a^*b^*$  plane, and Z is normal to that plane and coincides with c-axis. Referred to this set of coordinates, the thermal-expansion tensor is described by the expression:

$$\alpha_{ik} = \begin{vmatrix} a_{11} & a_{12} & a_{13} \\ a_{12} & a_{22} & a_{23} \\ a_{13} & a_{23} & a_{33} \end{vmatrix} \quad (1)$$

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the 6 subscripts in which are determined by the thermal expansions of the mentioned 6 interplanar spacings as follows:

$$\begin{aligned}
 \Delta_{100} &= a_{11}, \\
 \Delta_{010} &= a_{11} \cos^2 \gamma + a_{22} \sin^2 \gamma + 2a_{12} \cos \gamma \sin \gamma, \\
 \Delta_{110} &= a_{11} \cos^2 (a' H_{110}) + a_{22} \sin^2 (a' H_{110}) + 2a_{12} \sin (a' H_{110}) \cos (a' H_{110}), \\
 \Delta_{001} &= a_{11} \cos^2 \beta + a_{22} \cos^2 (c' Y) + a_{33} \cos^2 (c' H_{001}) + 2a_{12} \cos \beta \cos (b' Y) + \\
 &\quad + 2a_{13} \cos \beta \cos (c' c) + 2a_{23} \cos (c' Y) \cos (c' c), \\
 \Delta_{101} &= a_{11} \cos^2 (a' H_{101}) + a_{22} \cos^2 (H_{101} Y) + a_{33} \cos^2 (H_{101} c) + \\
 &\quad + 2a_{12} \cos (a' H_{101}) \cos (H_{101} Y) + 2a_{13} \cos (H_{101} a') \cos (H_{101} c) + \\
 &\quad + 2a_{23} \cos (H_{101} Y) \cos (H_{101} c), \\
 \Delta_{011} &= a_{11} \cos^2 (H_{011} a') + a_{22} \cos^2 (H_{011} Y) + a_{33} \cos^2 (H_{011} c) + \\
 &\quad + 2a_{12} \cos (H_{011} a') \cos (H_{011} Y) + 2a_{13} \cos (H_{011} a') \cos (H_{011} c) + \\
 &\quad + 2a_{23} \cos (H_{011} Y) \cos (H_{011} c).
 \end{aligned} \tag{5}$$

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$H_{hkl}$  means reciprocal-lattice vector;  $\alpha^*, \beta^*, \gamma^*$  and other angles can be expressed in terms of the crystallographic interaxial angles according to known equations. The transformation of the found tensor  $a_{ik}$  into the diagonal tensor  $a'_{ik}$ , i.e. reference of the tensor to the set of X', Y', Z' axes, is achieved using

$$D(\alpha) = -\alpha^3 + S_1\alpha^2 - S_2\alpha + S_3, \quad (6)$$

$$S_1 = a_{11} + a_{22} + a_{33},$$

$$S_2 = \begin{vmatrix} a_{11} & a_{12} \\ a_{12} & a_{22} \end{vmatrix} + \begin{vmatrix} a_{11} & a_{13} \\ a_{13} & a_{33} \end{vmatrix} + \begin{vmatrix} a_{22} & a_{23} \\ a_{23} & a_{33} \end{vmatrix},$$

$$S_3 = \begin{vmatrix} a_{11} & a_{12} & a_{13} \\ a_{12} & a_{22} & a_{23} \\ a_{13} & a_{23} & a_{33} \end{vmatrix}.$$

There are 3 figures; and 5 references, 4 Soviet, 1 U.K.  
The U.K. reference is: Y. A. Wooster, Textbook on  
Crystalphysics, Oxford, 1938.

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ASSOCIATION: Moscow State University imeni M. V. Lomonosov  
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SUBMITTED: May 21, 1959

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SOV/70-4-6-9/31

AUTHORS: Zhdanov, G. S., Solov'yev, S. P., Venevtsev, Yu. N.,  
Ivanova, V. V.

TITLE: Internal Fields in the Orthorhombic Modification of  
 $\text{BaTiO}_3$

PERIODICAL: Kristallografiya, 1959, Vol 4, N 6, pp 855-861 (USSR)

ABSTRACT: Internal fields in orthorhombic (pseudomonoclinic) barium titanate are computed according to ionic-displacement data reported in Phys. Rev., 105, 3, 856, 1957. Computations are based on the assumption that point ionic charges are displaced parallel to the edges of monoclinic unit cells, twice as small as orthorhombic cells, and are superposed by the similarly displaced point dipole moments. The latter's magnitude is determined as the product of ionic polarization and the affecting internal field. The known equation:

$$E_x = p_x \sum \frac{2x_i^2 - y_i^2 - z_i^2}{(x_i^2 + y_i^2 + z_i^2)^{3/2}} + p_y \sum \frac{3x_i y_i}{(x_i^2 + y_i^2 + z_i^2)^{3/2}} + p_z \sum \frac{3x_i z_i}{(x_i^2 + y_i^2 + z_i^2)^{3/2}}$$

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